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Question Paper Code

11549

B.E. / B.Tech. - DEGREE EXAMINATIONS, NOV/DEC 2022

Fifth Semester

Electronics and Instrumentation Engineering

(Common to Instrumentation and Control Engineering)

20EIPC501 - ANALYTICAL INSTRUMENTATION

(Regulations 2020)

Duration: 3 Hours

Max. Marks: 100

$PART - A (10 \times 2 = 20 Marks)$

Answer ALL Questions

1.	State beer-Lambert law.	Marks, K-Level, CO 2,K1,CO1
2.	List the different types of spectrophotometers.	2,K1,CO1
3.	Define Chromatography.	2,K1,CO2
4.	List the various components used in HPLC.	2,K1,CO2
5.	Discuss the advantages of Thermal Conductivity analyzer.	2,K2,CO3
6.	Mention the applications of thermal conductivity gas analyzer.	2,K2,CO3
7.	Define pH.	2,K1,CO4
8.	Differentiate between glass electrode and reference electrode.	2,K2,CO4
9.	State the basic principle of NMR.	2,K2,CO5
10.	Mention the advantages of Mass Spectrometry.	2,K2,CO5

PART - B $(5 \times 13 = 65 \text{ Marks})$

Answer ALL Questions

11. a) Explain the various types of sources and detectors used for UV-Visible ^{13, K2,CO1} Spectrophotometers.

OR

- b) Explain the working principle of FTIR Spectrometer with neat 13, K2,CO1 diagram.
- 12. a) Draw the schematic diagram of a gas chromatography& explain the ^{13, K2,CO2} components in detail.

OR

b) Draw & explain the instrumentation of High Pressure Liquid ^{13, K2,CO2} Chromatography in detail.

a) Explain in detail the working principle of O2 Analyzer based on 13, K2,CO3 13. magnetic susceptibility.

- b) Elucidate the working principle of hot wire Thermal Conductivity 13, K2,C03 Analyzer.
- a) List the types of electrodes used for pH measurement and explain the 13, K2,CO4 14. principle of pH measurement.

Explain the silica analyzer and sodium analyzer in detail.

13, K2,CO4

Explain the working principle of a pulsed Fourier Transform NMR 13, K2,C05 15. a) spectrometer with neat diagram

b) With neat Sketch explain the instrumentation setup of continuous wave 13, 1 NMR Spectrometer and its function.

PART - C $(1 \times 15 = 15 \text{ Marks})$

- 16. a) (i) Explain the constructional detail and working of glass electrode.
 - 8,K2,CO4
 - (ii) Explain about the working principle of quadrupole mass analyzers

7,K2,CO5

b) (i) Explain the principle of conductivity measurement in detail.

8,K2,CO4

(ii) Explain about the working principle of time of flight mass analyzers.

7,K2,CO5