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Question Paper Code	12549
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B.E. / B.Tech - DEGREE EXAMINATIONS, NOV / DEC 2023
 First Semester
 (Common to All Branches except Computer Science and Business Systems)
20BSCY101 - ENGINEERING CHEMISTRY
 (Regulations 2020)

Duration: 3 Hours

Max. Marks: 100

PART - A (10 × 2 = 20 Marks)

Answer ALL Questions

- | | <i>Marks,
K-Level, CO</i> |
|---|-------------------------------|
| 1. What is Calgon conditioning? How is it functioning in water treatment? | 2,K2,CO1 |
| 2. Define catalytic poisons. | 2,K1,CO1 |
| 3. What is an electrochemical series? | 2,K2,CO2 |
| 4. What is sacrificial anode? How does it protect a submerged pipeline? | 2,K2,CO2 |
| 5. Distinguish between coal and coke. | 2,K1,CO3 |
| 6. Define calorific value. | 2,K1,CO3 |
| 7. Define: Nuclear chain reaction. | 2,K1,CO4 |
| 8. Mention the advantages of lithium ion cells? | 2,K2,CO4 |
| 9. Why thermosetting plastics cannot be remolded? | 2,K2,CO5 |
| 10. Distinguish between bulk particles and nano-particles. | 2,K2,CO5 |

PART - B (5 × 13 = 65 Marks)

Answer ALL Questions

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| 11. a) | Explain the demineralization of water by the ion-exchange process. How are exhausted cation and anion exchange resins regenerated? | 13,K2,CO1 |
| | OR | |
| b) | Explain the role adsorbent in pollution abatement. | 13,K2,CO1 |
| 12. a) | (i) Derive Nernst equation for single electrode potential and give its significance. | 8,K2,CO2 |
| | (ii) Calculate the reduction potential of Cu/Cu ²⁺ (0.5 M) at 25°C. given that E° = 0.337: (Cu ²⁺ = 0.5M). | 5,K2,CO2 |
| | OR | |
| b) | What is electroplating? Discuss the plating composition, mechanism of copper plating. | 13,K2,CO2 |
| 13. a) | Explain flue gas analysis by ORSAT method with suitable diagrams. | 13,K2,CO3 |

OR

- b) Describe the Otto – Hoffman of coke manufacture and the recovery of various by-products. *13,K2,CO3*
14. a) What is a nuclear reactor? Describe the components of a light-water nuclear power plant with a suitable block diagram. *13,K3,CO4*

OR

- b) Explain the construction and working of Hydrogen-Oxygen fuel cells with a neat diagram. *13,K3,CO4*
15. a) Illustrate the preparation of the following polymers. *13,K2,CO5*
Nylon 6,6 b) Teflon c) Kevlar

OR

- b) Discuss the CVD and Laser ablation techniques for the synthesis of nanoparticles. *13,K2,CO5*

PART - C (1 × 15 = 15 Marks)

16. a) (i) How is the softening of water carried out using the zeolite process? *8,K2,CO1*
(ii) What is synthetic petrol? How is it manufactured by the Bergius process? *7,K2,CO3*

OR

- b) (i) What is desalination? Describe the reverse osmosis method with neat diagram. *8,K2,CO1*
(ii) Write notes on: Bio-diesel. *7,K2,CO3*