

Reg. No.

Question Paper Code

13757

B.E. / B.Tech. - DEGREE EXAMINATIONS, APRIL / MAY 2025

Second Semester

Mechanical Engineering

(Common to Mechanical and Automation Engineering)

24BSPH202 – PHYSICS OF MATERIALS

Regulations - 2024

Duration: 3 Hours

Max. Marks: 100

PART - A (MCQ) (10 × 1 = 10 Marks)

Answer ALL Questions

Marks *K-
Level* CO

- | | | | |
|--|---|----|-----|
| 1. The main use of a binary phase diagram in metallurgy is _____ | 1 | K1 | CO1 |
| (a) To determine the electrical conductivity of an alloy | | | |
| (b) To predict the phases present at specific temperatures and compositions | | | |
| (c) To measure the density of an alloy | | | |
| (d) To assess the color change during heating | | | |
| 2. If an alloy's composition lies at the midpoint of a tie line, the relative proportions of the two phases are: | 1 | K1 | CO1 |
| (a) 50% each | | | |
| (b) 25% and 75% | | | |
| (c) Dependent on the temperature | | | |
| (d) Not predictable | | | |
| 3. _____ is the predominant microstructure of plain carbon steel when slow-cooled? | 1 | K1 | CO2 |
| (a) Bainite | | | |
| (b) Pearlite and ferrite | | | |
| (c) Martensite | | | |
| (d) Cementite | | | |
| 4. Cementite in the microstructure of steel is: | 1 | K1 | CO2 |
| (a) A solid solution of carbon in iron | | | |
| (b) A stoichiometric compound of Fe ₃ C | | | |
| (c) An amorphous phase in steel | | | |
| (d) Pure carbon in crystalline form. | | | |
| 5. Which property is most characteristic of ferromagnetic materials? | 1 | K1 | CO3 |
| (a) High electrical conductivity | | | |
| (b) Spontaneous alignment of magnetic domains | | | |
| (c) Resistance to magnetization | | | |
| (d) Lack of response to external magnetic fields | | | |
| 6. Which material is commonly used in the cores of transformers? | 1 | K1 | CO3 |
| (a) Ferrites | | | |
| (b) Copper | | | |
| (c) Aluminum | | | |
| (d) Stainless steel | | | |
| 7. The dielectric constant (ϵ_r) of a material represents: | 1 | K1 | CO4 |
| (a) The material's resistance to magnetic fields | | | |
| (b) The ratio of permittivity of the material to free space | | | |
| (c) The electric conductivity of the material | | | |
| (d) The material's heat resistance | | | |
| 8. The BCS theory explains superconductivity as: | 1 | K1 | CO4 |
| (a) Magnetic domain alignment | | | |
| (b) Electron pairing into Cooper pairs | | | |
| (c) Ionization of atoms at low temperatures | | | |
| (d) High energy electron interactions | | | |
| 9. Nanomaterials are defined by: | 1 | K1 | CO5 |
| (a) Their molecular composition | | | |
| (b) Their size, which is typically less than 100 nanometers | | | |
| (c) Their high electrical resistance | | | |
| (d) Their use only in the automotive sector | | | |
| 10. What is the primary purpose of a drug delivery system? | 1 | K1 | CO6 |
| (a) To enhance the taste of medication | | | |
| (b) To deliver drugs to specific target sites in the body effectively | | | |
| (c) To reduce the cost of drug manufacturing | | | |
| (d) To improve the shelf life of medications | | | |

PART - B (12 × 2 = 24 Marks)

Answer ALL Questions

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|---|---|----|-----|
| 11. Write a short note on phase diagram. | 2 | K1 | CO1 |
| 12. List the differences between eutectic reaction and Peritectic reaction. | 2 | K2 | CO1 |
| 13. State Fick's first and second laws. | 2 | K1 | CO2 |
| 14. Mention methods of increasing fatigue life. | 2 | K1 | CO2 |
| 15. Define magnetic Permeability. | 2 | K1 | CO3 |
| 16. State Curie – Weiss law. | 2 | K1 | CO3 |
| 17. List any three uses of a dielectric material. | 2 | K1 | CO4 |
| 18. Define SQUID. | 2 | K1 | CO4 |
| 19. Define pseudoelastic effect. | 2 | K1 | CO5 |
| 20. What is the role of matrix and reinforcement in composites? | 2 | K2 | CO5 |
| 21. Write short notes on bottom up process. | 2 | K1 | CO6 |
| 22. Give any two examples of ceramic biomaterials. | 2 | K1 | CO6 |

PART - C (6 × 11 = 66 Marks)

Answer ALL Questions

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|---|----|----|-----|
| 23. a) Discuss about Gibbs free energy and its applications to alloy formation. | 11 | K2 | CO1 |
| OR | | | |
| b) Explain isomorphous system with one example. | 11 | K2 | CO1 |
| 24. a) Draw the iron-carbon phase diagram and explain the various regions in it. | 11 | K2 | CO2 |
| OR | | | |
| b) Explain the procedure of Brinell hardness test and give its advantages and limitations. | 11 | K2 | CO2 |
| 25. a) Distinguish between dia, para, and ferro magnetism. | 11 | K2 | CO3 |
| OR | | | |
| b) Explain the domain structure in ferromagnetic materials. Show how the Hysteresis curve is explained on the basis of domain theory. | 11 | K2 | CO3 |
| 26. a) Write in detail about the various factors contributing to breakdown in dielectrics. | 11 | K2 | CO4 |
| OR | | | |
| b) Discuss about the Type-I and Type-II Superconductors. | 11 | K2 | CO4 |
| 27. a) Summarize the structural, mechanical, electrical, magnetic and chemical properties of metallic glasses. | 11 | K2 | CO5 |
| OR | | | |
| b) Discuss about biomaterials and its applications. | 11 | K2 | CO5 |
| 28. a) Explain quantum confinement and quantum structures in nanomaterials. | 11 | K2 | CO6 |
| OR | | | |
| b) Discuss the properties and applications of nanomaterials in electronics and photonics. | 11 | K2 | CO6 |